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Hypervalent Tellurium Compounds Containing Te-N Interactions. Mononuclear and Polynuclear Derivatives

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Various organotellurium derivatives containing the pendant arm (2-Me₂NCH₂)C₆H₄ group have been prepared and characterized. The structure of a hydrolysis by-product, [(Ph₂TeCl)₂O]₂, was also established.

Keywords: tellurium; hypervalent compounds; phosphorus ligands

INTRODUCTION

The chemistry of organotellurium compounds containing intramolecular N→Te interactions received considerably interest in recent years due to their enhanced thermal and hydrolytic stability^[1,2]. In order to investigate the effect on the N→Te interaction of the ligand *trans* to nitrogen we decided to use dithio ligands. A few organotellurium compounds containing both *ortho*tellurated benzylamino and dithiocarbamato ligands have been

reported so far, e.g. RTeS(S)CNR'₂^[3] and RTe[S(S)CNEt₂]₃^[3,4], but no derivatives containing thiophosphorus ligands are described^[5].

Here we report on the synthesis and characterization of various $[2-(Me_2NCH_2)C_6H_4]TeS(S)PR_2$ [R = Me, Et, Ph,) as well as the molecular structure of $[2-(Me_2NCH_2)C_6H_4Te]_2$, $[(Ph_2TeCl)_2O]_2$ and $[2-(Me_2NCH_2)C_6H_4]TeS(S)PPh_2$.

RESULTS AND DISCUSSION

The diorganoditelluride, [2-(Me₂NCH₂)C₆H₄Te]₂, was prepared according to a literature method^[4] and its molecular structure was determined by X-ray diffractometry (Fig. 1). The crystal consists of monomeric dinuclear molecules which exhibit a T-shaped geometry around both tellurium atoms as result of the intramolecular N→Te [N^{...}Te 2.84, 2.90 Å] interactions. The cordination geometry around the Te atoms can be described as *pseudo*-trigonal bipyramidal, with N and the second Te atoms in axial positions [N^{...}Te-Te 166.2, 169.4°] and two lone pairs and the carbon atom in the equatorial plane.

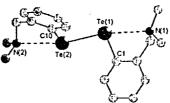


Fig. 1. View of the structure of [2-(Me₂NCH₂)C₆H₄Te]₂.

The reaction of Ph₂TeCl₂ and [2-(Me₂NCH₂)C₆H₄]Li affords

isolation of [2-(Me₂NCH₂)C₆H₄]TePh₂Cl. During evaporation of a CDCl₃/DMSO solution used for recording the NMR spectra, few crystals of a hydrolysis product, *i.e.* [(Ph₂TeCl)₂O.(CH₃)₂SO, were isolated. The crystal contains two similar independent molecules and the solvent molecules are disordered. The bridging nature of one of the Cl atoms from the monomeric unit [Cl(2)-Te(2) 2.815, Cl(2)-Te(1') 3.326 Å] results in a dimeric structure, [(Ph₂TeCl)₂O]₂ (Fig. 2), containing an eight-membered Te₄O₂Cl₂ inorganic ring. *Trans*-annular Te⁻⁻⁻Cl (3.59 Å) interactions are also present, leading to a condensed polycyclic system.

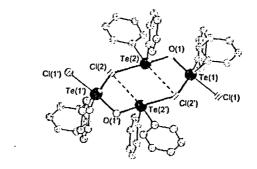


Fig. 2. View of the structure of $[(Ph_2TeCl)_2O]_2$.

The reaction between this diorganoditelluride and the corresponding disulfane, $[R_2P(S)S]_2$ (R = Me, Et, Ph) affords isolation of diorganodithiophosphinato derivatives:

$$[2-(Me2NCH2)C6H4Te]2 + [R2P(S)S]2 \longrightarrow$$

$$2 [2-(Me2NCH2)C6H4]TeS(S)PR2$$

All compounds were characterized by multinuclear NMR spectroscopy (¹H, ¹³C and ³¹P) and the molecular structure of

[2-(Me₂NCH₂)C₆H₄]TeS(S)PPh₂ has been determined (Fig. 3). The compound is almost T-shaped as a result of the N \rightarrow Te (2.439 Å) intramolecular interactions. The dithio ligand is monodentate [Te(1)-S(1) 2.5416 Å] and twisted to bring the non-bonded S(2) atom much closer to the tellurium atom [Te(1)-S(2) 3.878 Å].

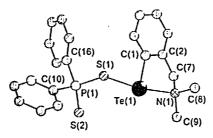


Fig. 3. View of the structure of [2-(Me₂NCH₂)C₆H₄]TeS(S)PPh₂.

References

- [1] N. Sudha and H. B. Singh, Coord. Chem. Rev., 135/136, 469 (1994).
- [2] Z. Majeed, W. R. McWhinnie and T. A. Hamor, J. Organometal. Chem., 549, 257 (1997).
- [3] H. B. Singh, N. Sudha, A. A. West and T. A. Hamor, J. Chem. Soc., Dalton Trans., 907, (1990).
- [4] H. B. Singh, N. Sudha and R. T. Butcher, *Inorg. Chem.*, 31, 1431, (1992).
- [5] I. Haiduc, R. B. King and M. G. Newton, Chem. Rev., 94, 301, (1994).
- [6] R. Kaur, H. B. Singh and R. T. Butcher, Organometallics, 14, 4755, (1995).